



Inorganic and Analytical Chemistry (higher education planning materials like oil and natural gas)

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Date :2007-08-01 1st edition 1st printing 2007-08-01 This textbook introduces the first distributed system from the macro (dilute solution. gel) and the fundamental nature of the basic principles of chemical reactions (energy changes in reaction rate. the reaction direction. the equilibrium move). and then introduced from the microscopic structure of matter (atoms. molecules. crystals) the basics. Then outlined the basics of quantitative chemical analysis. discusses the various types of chemical equilibrium in solution and titration analysis in the application. and the most commonly used instrument analysis made profile. Finally an important and complex element of the separation and enrichment material. Contents: Introduction Chapter I Section I of gas and solution gas dispersion system II III IV non-electrolyte solution. the concentration of dilute solution of colloidal solution according to the number of V VI VII polymer solution and emulsion section of the electrolyte solution thinking questions the basis of chemical thermodynamics chapter exercises and basic knowledge of chemical equilibrium thermodynamics Section II thermo-chemical reaction in the direction III IV chemical equilibrium problems thinking...

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